

Binding of Fatty Acids to β -Cryptogein: Quantitative Structure–Activity Relationships and Design of Selective Protein Mutants

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Received May 22, 2004

Binding of fatty acids to cryptogein, the proteinaceous elicitor from *Phytophthora*, was studied by using molecular docking and quantitative structure–activity relationships analysis. Fatty acids bind to the groove located inside the cavity of cryptogein. The structure–activity model was constructed for the set of 27 different saturated and unsaturated fatty acids explaining 87% (81% cross-validated) of the quantitative variance in their binding affinity. The difference in binding between saturated and unsaturated fatty acids was described in the model by three electronic descriptors: the energy of the lowest unoccupied molecular orbital, the energy of the highest occupied molecular orbital, and the heat of formation. The presence of double bonds in the ligand generally resulted in stronger binding. The difference in binding within the group of saturated fatty acids was explained by two steric descriptors, i.e., ellipsoidal volume and inertia moment of length, and one hydrophobicity descriptor, i.e., lipophilicity. The developed model predicted strong binding for two biologically important molecules, geranylgeranyol and farnesol playing an important role in plant signaling as lipid anchors of some membrane proteins. Elicitin mutants selectively binding only one type of ligand were designed for future experimental studies.

INTRODUCTION

Study of relationships between a pathogen and a plant is important for protection of plants against infections. Elicitors are the molecules participating in the plant–pathogen interactions. Secretion of elicitors by a pathogen to the surrounding environment and their interaction with plant cells can induce plant hypersensitive reaction. Elicitins make up the highly conserved family of protein elicitors.¹ They are secreted specifically by the fungi *Oomycete* genera *Pythium* and *Phytophthora*.^{2,3} The biological function of elicitors is currently unknown. The study of physiological effects on tobacco plants revealed that elicitors have the ability to induce so-called “systemic acquired resistance” against the pathogen attack accompanied by restricted leaf necrosis.^{3,4} The response is induced by the interaction of elicitors with a putative receptor located on the cytoplasmic membrane of tobacco cells⁵ composed of a calcium channel⁶ and a glycoprotein.⁷ The transfer of a signal through the receptor triggers phosphorylation–dephosphorylation cascades in the tobacco resulting in alkalization of the extracellular medium, efflux of potassium and chloride ions, influx of calcium, production of the active species from oxygen,^{5,8} and changes in the composition of the cell wall.⁹ The primary structure of mature elicitors is composed of 98 amino acids (10 kDa) that are interconnected by three disulfide bridges. Elicitins can be classified according to their *pI* as α -elicitors (*pI* < 7) and β -elicitors (*pI* > 7). The β -elicitors generally induce a greater necrotic effect than the α -elicitors¹⁰ due to

the presence of polar amino acids at necrotic sites located on the protein surface.^{11,12}

The three-dimensional structures of two elicitor family members, cryptogein and cinnamomin, were determined by X-ray crystallography.^{13,14} The structure of cryptogein was also studied by NMR spectroscopy.¹⁵ The structures of elicitors are composed of five α -helices and one β -sheet arranged in a unique protein fold. A hydrophobic cavity is located in the protein core and connected with the protein surface by a tunnel. The original proposal that elicitors may facilitate transfer of sterols¹⁶ was corroborated by the crystal structures of cryptogein in complex with dehydroergosterol (DHE)¹⁷ and cholesterol.¹⁸ Binding of ligands to the cavity seems to be essential for consecutive association of the elicitor with a receptor and induction of a biological response in a plant.¹⁹ Besides sterols, also fatty acids (FAs) can bind to the internal cavity of elicitors²⁰ making them functionally similar to the family of plant lipid transfer proteins. Interestingly, plant lipid transfer proteins can associate with the same receptor in tobacco as elicitors,²¹ and they can bind FAs and phospholipids but not sterols.

Here we study binding of FAs to the protein cryptogein at atomic detail. Models of the FA–cryptogein complexes were constructed by molecular docking to obtain information about position and conformation of ligands inside the cavity. These models consequently served for design of mutant proteins binding selectively one type of ligand. The physicochemical properties important for the binding were identified by quantitative structure–activity relationships (QSAR) analysis, and the model served for prediction of binding affinities of two lipids, farnesol and geranylgeranyol, that play an important role in plant signaling.²²

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METHODS

Modeling of Protein–Ligand Complexes. Protein–ligand complexes were constructed by the method of molecular docking using the program AUTODOCK 3.0.²³ The structure of protein cryptogein in the open state was obtained by withdrawal of DHE and water molecules from the crystal structure of the cryptogein-DHE complex solved to the resolution 2.15 Å (PDB-ID 1BXM). The open state corresponds to the structure with the first α -helix rearranged and with three side-chains of hydrophobic residues rotated outside the protein core thus creating a larger cavity for ligands. The hydrogen atoms were added by using the program WHATIF 5.0.²⁴ Energy maps were calculated by the AUTOGRID module of AUTODOCK with the size $100 \times 100 \times 100$ points and separation 0.25 Å. The structures of ligands (Table 1) were created in the program INSIGHT II (Accelrys, U.S.A.), geometries were energetically optimized by the program MOPAC 6.0 (QCPE, U.S.A.), and rotatable bonds were assigned using the AUTOTORS module of AUTODOCK. The carboxyl group of FAs was modeled as ionized according to pH 7.0 and used for experimental determination of the binding constants.²⁰ Docking itself was conducted by Lamarckian genetic algorithm using the following parameters: 50 runs for every docking, 50 individuals in a population, the maximal degree of evaluation 1.5×10^6 , the maximal number of generation 27 000, the value of elitism 1, the rate of mutation 0.02, and the rate of crossing-over 0.5. A local search was based on pseudo-Solis and Wets algorithm with the maximum of iterations 300. Resulting orientations from every docking run were divided into groups using cutoff root mean-squared deviation (RMSD) 0.5 Å.

Construction of QSAR Model. Molecular descriptors listed in Table 2 were calculated using the programs TSAR 3.1 and VAMP 6.0 (Accelrys, U.S.A.). The molecular structures were built in the program Insight II (Accelrys, U.S.A.), optimized in a vacuum using the AM1 method and the BFGS minimization algorithm and read to the spreadsheet of program TSAR. Alternatively, biologically relevant conformations were obtained from the molecular docking. TSAR calculates molecular descriptors, i.e., physicochemical properties, structural parameters, and topological indexes, in the batch mode using the algorithms described in the program manual. The quantum mechanical descriptors, i.e., energies of frontier molecular orbitals and dipole moments, were calculated using the semiempirical molecular orbital package VAMP interfaced by TSAR. AM1 Hamiltonian was used for these calculations. Partial Least Squares Projection to Latent Structures (PLS) was used for identification of variables important for an explanation of experimental data.²⁵ PLS is a multivariate projection method designed for extraction and visualization of hidden phenomena in the data by correlating principal components of an independent (X) and a dependent (Y) data matrix. In our case, the X matrix was represented by molecular descriptors and a y vector by experimentally determined binding affinity (Table 1). The data set used for development of the QSAR models reported in this study is available from the Internet at <http://ncbr.chemi.muni.cz/~jiri/resources.html#qsar>. Variable selection was done manually based on the visual inspection of scores, loadings, and variable importance in projection plots.²⁶ The statistical quality of the models was evaluated

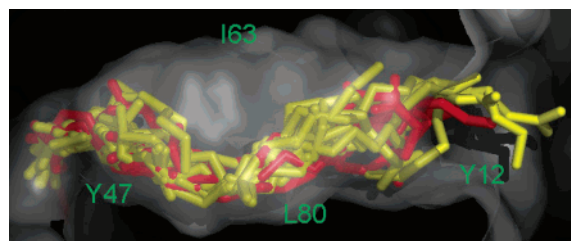


Figure 1. FA molecules docked to the cavity of cryptogein. Carbon chains align lengthwise with the groove located in the cavity. Saturated FAs (18 molecules) are in yellow, while unsaturated FAs (11 molecules) are in red. Active site is represented by the molecular surface with selected amino acids labeled. The hydrogen atoms are not shown for clarity.

by the correlation coefficient (R^2) and the cross-validated correlation coefficient (Q^2). The permutation test was conducted to access the risk of autocorrelation.

RESULTS

Molecular docking was used for the positioning of ligands to the internal cavity of cryptogein. Initially, the DHE molecule was removed from the cavity and docked back to the protein in order to optimize and validate the docking protocol. Docked orientation agreed well with the orientation from the crystal structure with RMSD equal to 1.4 Å. The hydrogen bond between DHE and Y47 was correctly identified by docking. All 29 FAs were docked to the cavity of cryptogein, and the quality of the fit was accessed by the scoring energies of the program AUTODOCK. The best orientations of the unsaturated FAs had the carbon chain positioned lengthwise of the groove located inside the protein cavity. Docking of long-chain saturated FAs provided two equally good orientations with ligand molecules bound in packed and extended conformation. The difference of the scoring energies between packed and extended conformation was negligible, but QSAR analysis revealed that only the extended conformation is biologically relevant. Furthermore, the similar binding mode for all studied ligands was obtained using extended conformations for long-chain saturated FAs (Figure 1) even though the superposition of the carboxylic group of all ligands could not be achieved due to the extreme length of some of the ligands compared to the size of the internal cavity of cryptogein. FAs bound to the cavity interact mainly with the residues making up the groove: Y33, P42, Y47, L41, V75, L80, L82, and Y87.

QSAR analysis was conducted for the data matrix of 29 ligands described by 28 molecular descriptors (X) and 1 experimentally determined binding affinity (y). Molecular descriptors calculated for the conformations of FAs obtained from the molecular docking, i.e., representing biologically relevant conformations, were used for the construction of QSAR models presented in the further text. Descriptors derived for the structures from docking were superior to the descriptors derived for the ground state conformations. This was confirmed by comparison of the statistical parameters of the QSAR models constructed for the two respective sets of descriptors (0.1 difference in R^2 and Q^2). The binding affinity describes the strength of ligand binding to the cryptogein and is defined as the ability of a ligand to displace DHE from the DHE-cryptogein complex. Briefly, the binding of DHE was measured as its fluorescence at 370 nm.

Table 1. List of Ligands

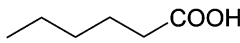
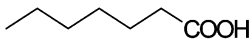
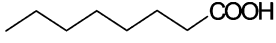
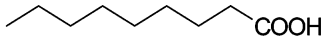
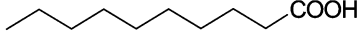
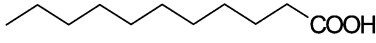
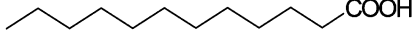
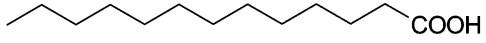
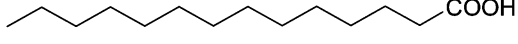
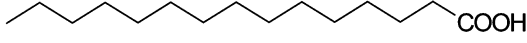
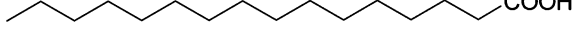
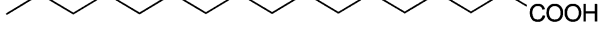
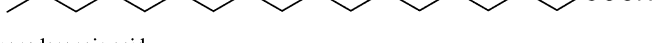





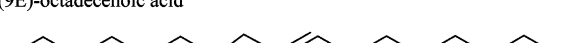

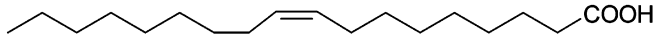
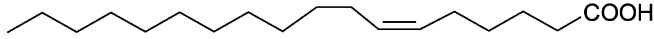
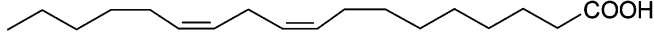
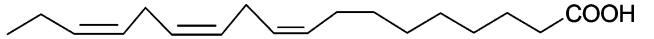
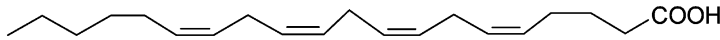
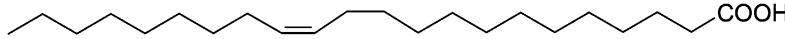
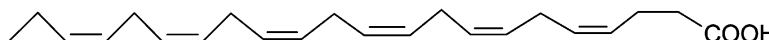
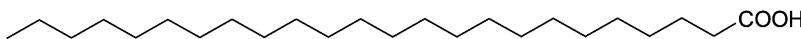
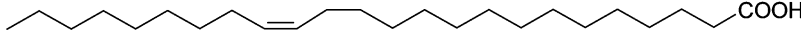
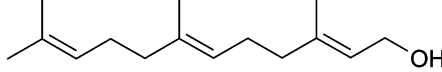
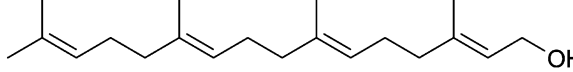
Systematic name and formula	Code	Bind. aff. ^a
hexanoic acid 	m6s	91.2
heptanoic acid 	m7s	73.6
octanoic acid 	m8s	93.8
nonanoic acid 	m9s	75.5
decanoic acid 	m10s	70.7
undecanoic acid 	m11s	75.5
dodecanoic acid 	m12s	75.6
tridecanoic acid 	m13s	64.7
tetradecanoic acid 	m14s	93.8
pentadecanoic acid 	m15s	83.9
hexadecanoic acid 	m16s	99.1
heptadecanoic acid 	m17s	92.4
octadecanoic acid 	m18s	108.1
nonadecanoic acid 	m19s	109.4
eicosanoic acid 	m20s	111.5
heneicosanoic acid 	m21s	106.3
docosanoic acid 	m22s	104.3
tricosanoic acid 	m23s	95.8
(9Z)-hexadecenoic acid 	m16n9c	47.0
(9E)-octadecenoic acid 	m18n9t	87.8

Table 1. List of Ligands

Systematic name and formula	Code	Bind. aff. ^a
(9Z)-octadecenoic acid 	m18n9c	29.5
(6Z)-octadecenoic acid 	m18n6c	60.1
(9Z,12Z)-octadecdienoic acid 	m18n912c	16.7
(9Z,12Z,15Z)-octadectrienoic acid 	m18n91215c	50.8
(5Z,8Z,11Z,14Z,-)-eicostetraenoic acid 	m20n581114c	34.6
(13Z)-docosenoic acid 	m22n13c	58.5
(4Z,7Z,10Z,13Z,16Z,19Z)-docosahexenoic acid 	m22n4710131619c	37.3
tetracosanoic acid 	m24s	101.8
(15Z)-tetracosanoic acid 	m24n15c	114.8
farnesol 	far	26.0 ^b
geranylgeranyl 	gerger	32.2 ^b

^a Relative binding affinity of FAs to cryptogein. Data from ref 20. ^b Values predicted from the model M3.

Cryptogein (250 nM) and DHE (100 nM) were incubated together in a buffer, and fluorescence of the complex was recorded. Thereafter the FA (18 μ M) was added. The binding affinity of a FA was expressed as a percentage of the fluorescence of the cryptogein-DHE complex and was inversely proportional to the fluorescence in the presence of FA (data adopted from publication²⁰). The smaller value of fluorescence corresponds to the higher strength of FA binding to the cryptogein. The initial QSAR model (M1) was developed for all FAs and all descriptors (Table 3). This model showed moderate statistical quality but assisted in identification of two outliers: (9Z)-octadecenoic acid (m18n9c) and (9Z,12Z)-octadecdienic acid (m18n912c). These two molecules showed the strongest binding to the cryptogein in the experiment. However, experimental values did not correspond to the scoring energies obtained from the docking, which showed good correlation with binding affinity for all other molecules in the data set (not shown). Either m18n9c and m18n912c bind to the cavity by a different binding mode compared to other molecules or their binding affinities were overestimated in the experiment. The new QSAR model

(M2) has been derived for 27 FAs and 28 molecular descriptors. The large difference between R^2 and Q^2 indicates the presence of redundant variables, which were removed in the final (M3) model (Table 3). The final model shows good fit ($R^2 = 0.87$) and good internal predictive ability ($Q^2 = 0.81$). The model satisfactorily explains the difference in binding of saturated and unsaturated FAs (Figure 2). The variance between these two subgroups is described by the electronic parameters: the energy of the lowest unoccupied molecular orbital (LUMO), the energy of the highest occupied molecular orbital (HOMO), and the heat of formation (HF). We believe that energies of the frontier orbitals and heat of formation are not related to the binding process directly, but they capture different conformational behavior of FAs originating from the presence or absence of double bonds in their structures. The double bonds restrict the conformational freedom of unsaturated FAs compared to their saturated analogues and keep the molecules in the extended conformation needed for their binding to the internal cavity of elicitin (Figure 1). The remaining variance within the group of saturated FAs is explained by two steric

Table 2. List of Molecular Descriptors

code	descriptor	program
MM	molecular mass	TSAR 3.1
SA	size of molecular surface	TSAR 3.1
MV	size of molecular volume	TSAR 3.1
M1s	inertia moment 1—size	TSAR 3.1
M2s	inertia moment 2—size	TSAR 3.1
M3s	inertia moment 3—size	TSAR 3.1
M1l	inertia moment 1—length	TSAR 3.1
M2l	inertia moment 2—length	TSAR 3.1
M3l	inertia moment 3—length	TSAR 3.1
EV	size of ellipsoidal volume	TSAR 3.1
Log P	octanol—water partition coeff.	TSAR 3.1
LIP	lipophilicity	TSAR 3.1
LIPx	X lipophilic component	TSAR 3.1
LIPy	Y lipophilic component	TSAR 3.1
LIPz	Z lipophilic component	TSAR 3.1
MR	molecular refractivity	TSAR 3.1
TE	total energy	VAMP 6.0
EE	electronic energy	VAMP 6.0
NE	nuclear energy	VAMP 6.0
POL	mean polarizability	VAMP 6.0
HF	heat of formation	VAMP 6.0
LUMO	energy LUMO	VAMP 6.0
HOMO	energy HOMO	VAMP 6.0
DIPv	dipole moment	VAMP 6.0
DIPxv	X component dipole moment	VAMP 6.0
DIPyv	Y component dipole moment	VAMP 6.0
DIPzv	Z component dipole moment	VAMP 6.0

Table 3. List of QSAR Models

model	M1	M2	M3
number of objects (n)	29	27 ^a	27 ^a
number of descriptors (X)	28	28	6 ^b
number of components (A)	2	2	3
correlation coeff. (R^2)	0.58	0.79	0.87
cross-validated correlation coeff. (Q^2)	0.34	0.68	0.81

^a Excluded objects m18n9c and m18n912c. ^b Remaining variables M1l, EV, LIP, HF, LUMO, and HOMO.

and one hydrophobicity descriptor: ellipsoidal volume (EV), inertia moment of length (M1l), and lipophilicity (LIP), respectively.

Geranylgeranyl and farnesol resemble FAs in the length of nonbranched hydrophobic chain and in the terminal polar group. The QSAR model could be therefore used for the prediction of their binding affinity. Molecular descriptors were calculated for the structures in their docked conformations. The predicted values of relative binding affinity for farnesol and geranylgeranyl are 26.0 and 32.6, respectively.

DISCUSSION

Biological activity of elicitors is correlated with their ability to bind sterol.¹⁹ It is probably the elicitor-sterol complex that binds to the receptor located on the cytoplasmic membrane and induces a physiological response in tobacco. The sterol molecule bound inside the cavity may induce conformational changes necessary for the binding of elicitor to the receptor and triggering the plant defense reaction.¹⁹ Cryptogein is a nonspecific sterol carrier protein able to bind sterols and FAs.²⁰ Unlike sterols, free FAs are not present in noticeable amounts in membranes but can be liberated by the action of phospholipases A₁ and A₂. Plant lipid transfer proteins (LTPs) are able to bind and transfer FAs and phospholipids but not sterols.²¹ Among them, LTPs1 are constitutively expressed in specific tissues and/or induced as a response to stress. LTPs1 and elicitors share some structural and functional properties and compete with elicitors in binding to the same membrane receptors in tobacco.²⁷ The study of the other ligands that bind to the elicitors and LTPs and induce similar conformational changes as sterols is important for understanding of the elicitor-receptor interactions. The relationship between the complexation of elicitor with sterol and FA, respectively, is currently unknown and the design of protein mutants selectively binding either molecule can stimulate future research.

Computer modeling was used here to analyze complexation of elicitor β -cryptogein with FAs, geranylgeranyl, and farnesol, respectively. The binding mode of 29 different FAs, geranylgeranyl, and farnesol was predicted by using molecular docking. FAs bind to the groove located inside the

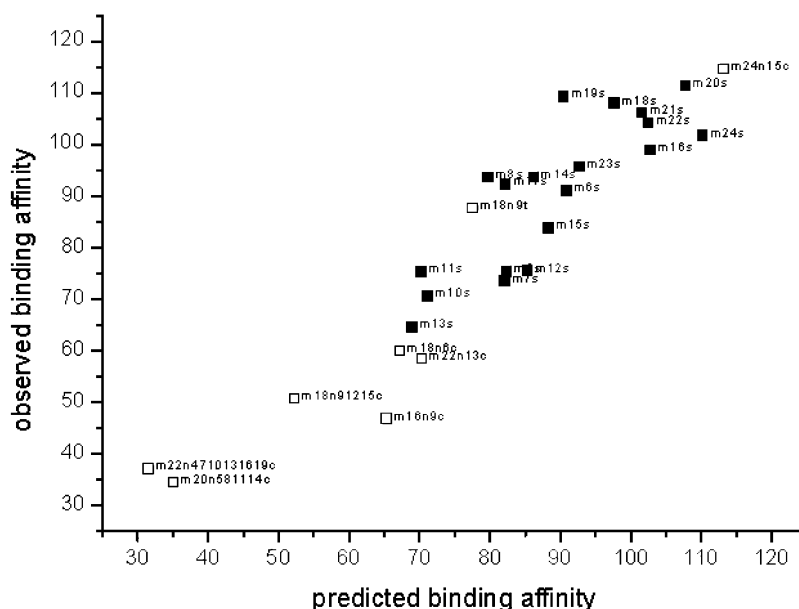


Figure 2. Plot of experimentally observed versus predicted binding affinities. Predictions were made using the QSAR model M3. Codes for FAs are provided in Table 1. Saturated FAs are represented by the filled squares, while unsaturated FAs are represented by empty squares.

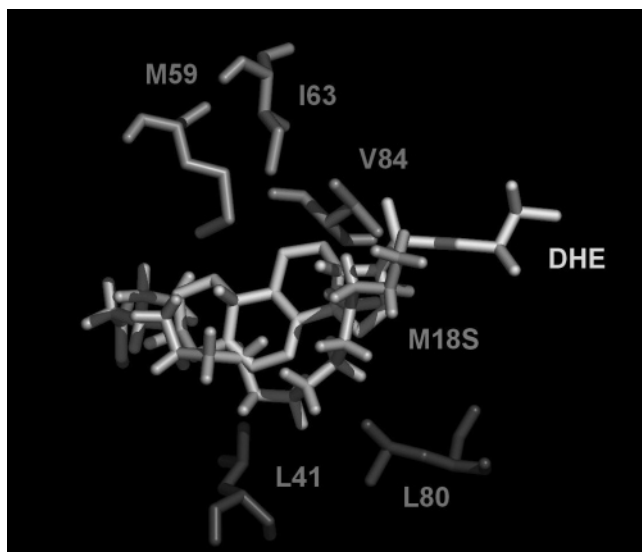


Figure 3. Amino acid residues suitable for substitutions leading to the protein variants with selective binding of DHE or FA. Substitutions of the residue M59, I63, and V84 by bulky hydrophobic amino acids will result in cryptogein preferentially binding FAs, while the same substitutions of the residues L41 and L80 will provide proteins preferring DHE. Octadecanoic acid (m18s) is displayed as the representative FA. Carboxylic group octadecanoic acid aligns with the alcoholic group of DHE.

protein cavity (Figure 1). Geranylgeranyol and farnesol are widespread in plant and animal cells in a form of prenylated proteins. These covalently attached lipids are recognized as being critically important for cellular signaling processes. Here we show that geranylgeranyol and farnesol bind to elicitors as efficiently as FAs. The importance of these interactions with elicitors or LTPs for the cell signaling should be tested. Geranylgeranyol and farnesol resemble FAs by their hydrophobic carbon chain as well as a terminally positioned polar group and can bind to the cryptogein by the same mode. QSAR analysis revealed that the binding of unsaturated ligands to the cavity of cryptogein is determined by their size and hydrophobicity. The strength of binding predicted for geranylgeranyol and farnesol from the QSAR model is in the same range as observed with the most strongly binding FAs. This finding may have important biological implications.

Structural alignment of the DHE-cryptogein and FA-cryptogein complexes revealed a different location of the ligands within the cavity (Figure 3). The higher flexibility of FAs allows them to accommodate the active site of cryptogein better than the more rigid DHE. We propose that FAs will not induce conformational changes in the protein structure, e.g. repositioning of the omega loop, observed in the DHE-cryptogein complex.¹⁷ This may have important implications for in/ability of the FA-ergosterol complexes to induce systemic acquired resistance. Comparison of DHE-cryptogein and FA-cryptogein complexes enabled us to design of mutations favoring one type of ligand molecule only. Substitution of the residues M59, I63, or V84 by a large hydrophobic amino acid, e.g. phenylalanine or tryptophan, should reduce binding of sterol to the cavity of cryptogein. These mutations should not have an effect on the binding of FAs, farnesol, and geranylgeranyol, filling a free space next to the molecules bound in the groove. Substitution in the position I63 is the most suited for this

purpose, because it is located right to the groove of protein. Substitutions of the amino acid residues L41 and L80 for larger hydrophobic amino acids should distinctively decrease binding of FAs, farnesol, and geranylgeranyol in the groove of the cavity, while preserving the binding of sterol. Characterization of protein mutants showing selective binding is currently ongoing in our laboratory.

ACKNOWLEDGMENT

This project was financially supported by the Czech Grant Agency (522/02/0925). Financial support is gratefully acknowledged.

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CI049832X